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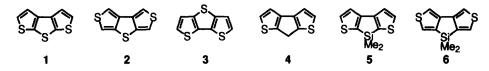
## SYNTHESIS OF DITHIENOTHIOPHENES, CYCLOPENTADITHIOPHENE AND SILACYCLOPENTADITHIOPHENES USING PALLADIUM-CATALYZED CYCLIZATION

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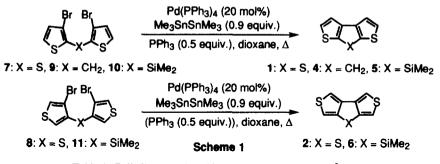
Abstract: The intramolecular cyclization of bromothiophene derivatives with hexamethylditin in the presence of tetrakis(triphenylphosphine)palladium(0) gave dithienothiophenes, cyclopentadithiophene, and silacvclopentadithiophenes in moderate to good yields. © 1997 Elsevier Science Ltd.

Dithienothiophenes (1, 2 and 3) have received considerable attention, because dithienothiophenes possess three different types of  $\pi$ -conjugations which play an important role as a spacer and donor,<sup>1</sup> and because these compounds give conjugated polymers with unique features.<sup>2</sup> Although [3,2-b: 2',3'-d]dithienothiophene (3) can be easily prepared and has been employed frequently,<sup>3</sup> the synthetic inconvenience of [2,3-b:3',2'-d]- and [3,4-b:3',4'-d]dithienothiophenes (1 and 2) and related compounds (4-6) has prevented the use of these compounds as building blocks for organic synthesis. We report here a convenient method for the synthesis of dithienothiophenes, cyclopentadithiophene and silacyclopentadithiophenes (1, 2 and 4-6).



In order to develop a new and general method for the synthesis of 1, 2 and 4-6, we investigated transition metal-catalyzed intramolecular cyclization reactions. The new method for the preparation of these compounds is outlined in Scheme 1 and Table  $1.^4$  The palladium-catalyzed cyclization of 7 was carried out using 20 mol% of Pd(PPh<sub>3</sub>)<sub>4</sub>, 0.9 equiv. of hexamethylditin, and 0.5 equiv. of triphenylphosphine in dioxane at 160 °C for 24 h to produce 1 in 75% yield (Entry 3). The reaction produced 1 in 55% yield at 110 °C for 24 h, but 38% of 7 was recovered (Entry 1). The reaction required 20 mol% of Pd(PPh<sub>3</sub>)<sub>4</sub>, and a similar reaction using 15 mol% of Pd(PPh<sub>3</sub>)<sub>4</sub> gave the product 1 in 67% yield (Entry 2). On the basis of these results, the palladium-catalyzed cyclization of 8 was carried out with Me<sub>3</sub>SnSnMe<sub>3</sub> (0.9 equiv.) in the presence of 20 mol% of Pd(PPh<sub>3</sub>)<sub>4</sub> [160 °C, 24 h] gave 4 only in 20% yield, but a lower temperature and longer time [110 °C, 96 h] enhanced the cyclization (Entry 5). Interestingly, the cyclization of the silicone derivatives  $10^5$  and  $11^5$  proceeded smoothly under palladium-catalyzed conditions to produce the corresponding silole

derivatives 5 and 6 in good yields (Entry 6 and 7).<sup>6</sup> In the absence of PPh<sub>3</sub>, the reaction of 11 produced 6 in a better yield (Entry 7). The oxidative addition of the C-Br bonds in 7-11 to the Pd(0) complex, followed by transmetallation and reductive elimination, produces monostannylated compounds which lead to the cyclization products by palladium-catalyzed intramolecular cross-coupling reaction.



Entry	Compound	Pd(PPh3)4 /mol%	Temp./C	Time/h	Product	Yield/%
1 <sup>b</sup>	7	15	110	24	1	55
2	7	15	160	24	1	67
3	7	20	160	24	1	75 <sup>c</sup>
4	8	20	160	72	2	78 <sup>c</sup>
5	9	20	110	96	4	50 <sup>c</sup>
6	10	20	160	72	5	72 <sup>d</sup>
7 <sup>b</sup>	11	20	160	72	6	64 <sup>c</sup>

Table 1. Palladium-catalyzed intramolecular cyclization.<sup>a</sup>

<sup>a</sup>A solution of the dibromide (0.25 mmol), Me<sub>3</sub>SnSnMe<sub>3</sub> (0.225 mmol), Pd(PPh<sub>3</sub>)<sub>4</sub> (0.05 mmol), and PPh<sub>3</sub> (0.125 mmol) in dioxane (2.5 ml) was heated in a sealed tube. The yields were determined by gas-chromatographic analysis. <sup>b</sup>In the absence of PPh<sub>3</sub>, <sup>c</sup>Isolated yields. <sup>d</sup>The isolated yield is 69%.

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## **References and Notes**

- a) Yui, K.; Ishida, H.; Aso, Y.; Otsubo, T.; Ogura, F. Chem. Lett. 1987, 2339; Mazaki, Y.; Hayashi, N.; Kobayashi, K. J. Chem. Soc., Chem. Commun. 1992, 1381; Brisset, H.; Thobie-Gautier, C.; Jubault, M.; Gorgnes, A.; Roncali, J. *ibid.* 1994, 1765. b) Hayashi, N.; Mazaki, Y.; Kobayashi, K. Tetrahedron Lett. 1994, 35, 5883.
- 2) Taliani, C.; Ruani, G.; Zanboni, R. Synth. Met. 1989, 28, C 507; Siekierski, M.; Przyluski, J.; Plocharski, J. *ibid.* 1993, 61, 217; Catellani, M.; Caronna, T.; Meille, S. V. J. Chem. Soc., Chem. Commun. 1994, 1911.
- 3) de Jong, F.; Janssen, M. J. J. Org. Chem. 1971, 36, 1645.
- Kelly, T. R.; Li, Q.; Bhushan, V. Tetrahedron Lett. 1991, 31, 161; Mori, M.; Kaneta, N.; Shibasaki, M. J. Org. Chem. 1991, 56, 3486, and references cited therein.
- 5) The compounds 10 and 11 were prepared in 66 and 60% yields, respectively, by the reactions of 2,3- and 3,4-dibromothiophenes with Bu<sup>n</sup>Li (1 eq.), followed by treatment with Me<sub>2</sub>SiCl<sub>2</sub> (0.5 equiv.).
- 6) The structures of all new compounds reported here were fully characterized by spectroscopic analyses. The selected data are as follows: 5: colorless oil, bp 84-86 °C/2.5 torr, MS (m/z) 222 (M<sup>+</sup>), 208, 147; <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>) δ 7.50 (d, J = 4.9 Hz, 2H), 7.12 (d, J = 4.9, 2H), 0.80 (s, 6H); <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>) δ 132.7, 132.0, 131.2, 118.5, -1.0. 6: colorless cryst., mp 103-104 °C, MS (m/z) 222 (M<sup>+</sup>), 208, 147; <sup>1</sup>H NMR (CD<sub>2</sub>Cl<sub>2</sub>) δ 7.64 (d, J = 2.4 Hz, 2H), 7.54 (d, J = 2.4, 2H), 0.41 (s, 6H); <sup>13</sup>C NMR (CD<sub>2</sub>Cl<sub>2</sub>) δ 148.2, 147.8, 131.8, 116.1, -1.1.

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